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4. Wind D. When two objects are at the same temperature. 4. Solid D. Allows convection; can be a good conductor of heat. 5. Liquid E. Any material that resists the movement of heat through it. 6. Gas F. No convection can occur in this. 2. Thermal Equilibrium 5. Convection 6. Thermodynamics E. The study of how heat moves. F.

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4. A 5 kg object is lifted up to the back of a 2 m tall ledge by a 20 N force pushing up a 10 m long ramp. A) Calculate the work in. B) Calculate the work out (what you got out). C) Calculate the efficiency of moving the object. D) Where did the extra energy go? 5. A 4 kg object starts at rest. A 25 N force pushes on it until it is going 5 m/s.

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Name: HW Unit 8:4 - cstephenmurray.com

HW Unit 8:6 — Thermodynamics Mr. Murray, IPC

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A. ___ You pick up a hot piece of metal and get burned. B. ___
You put your hand above a pan of hot water. C. ___ You feel the
heat from a brick wall when you put you hand next to the wall,
but not touching it. D.

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4. Static electricity 5. Electrically neutral A. A unit in measuring
the amount of charge B. The pushes and pulls that electric
charges exert on each other C. Property of matter responsible
for electrical events; it has two forms, positive and negative. D.
An object that has equal amounts of positive and negative
charges. E.

Electricity and Thermodynamics Review - cstephenmurray.com

HW Unit 8:5 — More Conservation of Energy Mr. Murray, IPC
cstephenmurray.com Match the following situations with their
Conservation of Energy Equations. 1. An object falls without
friction. 2. An object falls part way down. 3. An object at rest is
pushed. 4. A moving object is slowed down. 5. A moving object is
stopped. 6.

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HW Unit 8:5 — More Conservation of ... - cstephenmurray.com

HW Unit 8:1 — Guided Reading Mr. Murray, IPC

cstephenmurray.com Find the chapter on Work and Energy 1. Give the formula for work: 2. What are the units for work? 3. (Looking on the side). Work is how forces transfer e____. 4. If you push on something and it doesn't move, did you do work? 5. Why? 6.

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Example 1: A 2 kg mass going 1 m/s is pulled by an 8 N force for 4 sec. How fast is the mass going afterwards? +W Before $m = 2$ kg After 8 N 4 sec $m = 2$ kg $v = 1$ m/s $v = ?$ There is only one object, so $\Sigma p_{\text{before}} = mv$ and there is an external impulse. $34 = 2v$ $v_A = 17$ m/s $2(1) + 8(4) = 2(v_A) 2 + 32 = 2v$ $\Sigma p_{\text{before}} \pm I = \Sigma p_{\text{after}}$ $mv_B + Ft = mv_A$

The Law of Conservation of Momentum - Akers Physics

Unit 8: Temperature, Heat, & Thermodynamics. Conceptual - Wednesday 4/24. Physics Fun Fact! Warm Up Problem #1 & #2. Discuss pre-lab topics. Penny Lab! Friday 4/26: Finish Unit 8/Quiz. Academic - Tuesday 4/23. Happy Earth Day! - Video? Begin Unit 8 - Heat/Temperature . Unit 8 Notes. Temperature Formative (In class) Heat Practice ...

Unit 8: Temperature, Heat, & Thermodynamics

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• The second law of thermodynamics introduces the notion of entropy (S), a measure of system disorder (messiness) • U is the quantity of a system's energy, S is the quality of a system's energy. • Another C.P. Snow expression: - not knowing the 2nd law of thermodynamics is the cultural equivalent to never having read Shakespeare

Laws of Thermodynamics - MIT Haystack Observatory

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1+ ammonium NH_4^+ + 1- acetate $\text{C}_2\text{H}_3\text{O}_2^-$ 2- carbonate CO_3^{2-}
2-2- chromate CrO_4^{2-} 4 2-1- hydrogen carbonate HCO_3^- 3 1-1+ hydronium H_3O^+ + 1- hydroxide OH^- 1-1- nitrate NO_3^- 3 1-2- peroxide O_2^{2-} 2-3- phosphate PO_4^{3-} 3-2- sulfate SO_4^{2-} 4 2-2- sulfite SO_3^{2-} 3 2-Greek Prefixes Mono - 1 Di - 2 Tri - 3 Tetra - 4 Penta - 5 Hexa - 6 Hepta - 7 Octa - 8 Nona - 9

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Thermodynamics - Wikipedia

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