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1. increasing the surface area. 2. increasing the temperature. 3. increasing the strength of the intermolecular forces. 1 and 2 only. Liquids that have high vapor pressure and low boiling points are called. volatile liquids. The amount of heat required to melt one mole of a solid is called the: heat of fusion.

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Calculate the molality of a solution that 1.35 M and has a density

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of 1.28 g/mL. It is also known that the molar mass of the solute is 100.0 g/mole. 1 Answer. What is the molarity of 300g NaCl per litre solution. 1 Answer.

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Chemistry Class 12 Chapter 12 Aldehydes, Ketones, and Carboxylic Acid This chapter includes intext questions, exercises and two questionnaires. Questions related to structural formulas, names of four prospective condensations, chemical tests to differentiate terms of a compound, etc. are asked in the chapter.

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